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### Sciences

### Synthesis and Evaluation Of Antimicrobial, Antifungal Activity And Cell Viability of Tetrakis-[(4-methoxy)-phthalocyaninato] Zn (II).

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#### ABSTRACT

The cytotoxicity and antibacterial activity of Tetrakis-[(4-methoxy)-phthalocyaninato]Zn(II) were assayed by micro-dilution bioassay and the disc diffusion method against bacterial and fungal strains. All bacterial strains were particularly sensitive to the ZnPc with low MICs values ranging from 1 to 2 mg/mL and less active against the four *Candida* species tested with a diameter of growth inhibition about 6 mm and a MICs values ranging from 1 to 4 mg/mL. The cytotoxicity of our compound was evaluated on HEp-2 cells using the MTT salt reduction assay. The tested compound has no toxicity on HEp-2 cells ( $CC_{50} > 1000 \ \mu g. \ mL^{-1}$ ) and appears to be a promising antimicrobial agent.

Keywords: Anti-bacterial, Zinc phthalocyanine, Antifungal, Cytotoxicity,



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#### INTRODUCTION

Since their early discovery in the last century, phthalocyanines (Pcs) have been of great interest to chemists, physicists and industrial scientists and have become one of the most studied of all organic functional materials. In addition to their use as blue and green colorants, they are a versatile class of functional dyes, phthalocyanines have been widely studied for their applications in various disciplines. Phthalocyanines (Pcs) have been studied for many years and they are still the subject of intense investigations, active research fields have included their use for fuel cells, photovoltaic and solar cells [1,2,3], photosensitizers [4], gas sensors [5]...

For all of the above mentioned industrial applications, the growing antibacterial resistance among pathogens remains a global threat to health care as an endemic which keeps reminding scientific societies to develop new antibiotics. The interest towards the development of conventional antibiotics is reduced due to continual problems of bacterial resistance. Therefore, it is an emerging interest in the design and development of new materials with a novel and distinct mechanism in killing pathogens by replacing the conventional mechanisms.

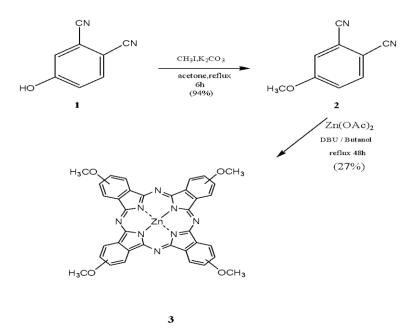
The present paper describes the synthesis of a substituted Zn-phthalocyanine and its structural characterization was done by NMR and mass spectrometry. The target phtalocyanine complex was tested against the antibacterial activity which was carried out on four Gram positive and four Gram negative bacteria: *Staphylococcus aureus* ATCC 25923, *Bacillus cereus* ATCC 11778, *Vibrio parahaemolyticus* CECT 511, *Listeria monocytogenes* ATCC19115, *Aeromonas hydrophila* ATCC 7966<sup>T</sup>, *Pseudomonas aeruginosa* ATCC 27853, *Salmonella typhimurium* ATCC 1408, and *Escherichia coli* ATCC 35218. On the other hand, the antifungal effect was tested against four *Candida* species (*C. albicans*, *C. parapsilosis*, *C. glabrata*, and *C. tropicalis*).

The determination of antibacterial activity was done using two methods, the disc diffusion assay and the micro-dilution assay. Additionally, the cytotoxicity of our compound was been studied in vitro with the aim to determine its safety for an eventual use as antifungal and antibacterial agent.

#### MATERIAL AND METHODS

UV-Vis spectra were recorded using a spectrophotometer UV-1600PC. NMR spectra were recorded with a Bruker AC-300 instrument. Column chromatography was performed with Merck 60 (230-400 mesh, 60 Å) silica gel. All solvents and reagents were purchased from Sigma-Aldrich, Acros and were used without further purification.

#### Synthesis



Scheme 1: Synthetic pathway for the synthesis of ZnPc.

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#### Synthesis of 1,2-Dicyano-4-(methoxy) benzene 2

To a solution of 4-hydroxyphtalonitrile **1** (150 mg, 1 mmol) in dry acetone (15 mL) was added  $K_2CO_3$  (158 mg, 1.1 mmol) with efficient stirring. 1 mL of iodomethane was then added and the mixture was stirred under reflux for 6 hours. The resulting mixture was cooled and  $K_2CO_3$  was collected by filtration. The solvent was removed under reduced pressure to give 154 mg of compound **2**.

Pale-yellow solid, m.p = 206-208 °C, <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>):  $\delta$  (ppm): 3.94 (s, 3H, CH<sub>3</sub>), 7.19 (dd, J<sub>1</sub> = 2.7 Hz, J<sub>2</sub> = 8.7 Hz, 1H), 7.28 (d, J = 2.7 Hz, 1H), 7.72 (d, J = 8.7 Hz, 1H). <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>):  $\delta$  (ppm): 56.22, 107.37, 115.23, 115.64, 117.49, 118.97, 119.16, 135.21, 162.66.

#### Tetrakis- [(4-methoxy)-phthalocyaninato]Zn(II) 3

A mixture of methoxyphthalonitrile **2** (100 mg, 0.6 mmol),  $Zn(OAc)_2$  (28 mg, 0.15 mmol), DBU (0.1 mL) and butanol (7 mL) was heated at reflux, under N<sub>2</sub> atmosphere, for 48 hours. Cooling to room temperature was followed by precipitation in methanol to yield 120 mg of ZnPc **3**.

Dark green solid, <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ ):  $\delta$  (ppm): 3.76 (br, 12H, OCH<sub>3</sub>), 7.32-8.40 (m, 12H).<sup>13</sup>C NMR (75 MHz, DMSO- $d_6$ ):  $\delta$  (ppm): 158.9, 131.9, 124.2, 118.9, 76.7, 60.3, 54.8. Anal. Calcd. for C<sub>36</sub>H<sub>24</sub>N<sub>8</sub>O<sub>4</sub>Zn: C, 61.95%; H, 3.47%; N, 16.05%. Found: C, 61.90%; H, 3.55%; N, 16.15%. MS (ESI): m/z calcd for C<sub>36</sub>H<sub>25</sub>N<sub>8</sub>O<sub>4</sub>Zn: 699.02 [M + H]<sup>+</sup>.

#### **Microbiological assays**

Two evaluated screening methods were used to analyse the antimicrobial activity of ZnPc including the disc diffusion assay (for the determination of the growth inhibition zone) and the micro-dilution assay (for the determination of the minimal inhibition concentration and minimal bactericidal/fungicidal concentration). The antimicrobial effect of synthetic Tetrakis-[(4-methoxy)-phthalocyaninato]Zn(II) was carried out on four Gram positive and four Gram negative bacteria: Staphylococcus aureus ATCC 25923, Bacillus cereus ATCC 11778, Vibrio parahaemolyticus CECT 511, Listeria monocytogenes ATCC 19115, Aeromonas hydrophila ATCC 7966T, Pseudomonas aeruginosa ATCC 27853, Salmonella typhimurium ATCC 1408, and Escherichia coliATCC 35218. On the other hand, the antifungal effect was tested against four *Candida* species (*C. albicans, C. parapsilosis, C. glabrata*, and *C. tropicalis*).

For the disc-diffusion assay, 10  $\mu$ L from the microorganisms culture stock were activated on a tube containing 9 mL of Mueller-Hinton (MH) broth (for bacteria) and Sabouraud (SB) chloramphenicol broth (for *Candida* strains), then incubated at 37 °C during 24 h. After one night, the cultures were used and the optical density was adjusted to 0.1 at OD<sub>600</sub> for bacteria and 0.4 at OD<sub>540</sub> for *Candida* strains corresponding approximately to 10<sup>7</sup> UFC/mL. The inoculums of bacteria and yeasts were streaked onto MH and SB agar plates, respectively, using a sterile cotton swab, as recommended by the CA-SFM EUCAST 2017. Sterile filter discs (diameter 6 mm, Biolife Italy) were impregnated with 20  $\mu$ L of the ZnPc sample (32 mg/mL) and placed on the appropriate agar media. Ampicillin (10 mg/mL; 10  $\mu$ L/disc) and Amphotericin B (10 mg/mL; 10  $\mu$ L/disc) were served as positive standards. After 24 h of culture (37 °C), the diameter of the growth inhibition zone (IZ) was measured using a flat rule. Each experiment was carried out in triplicate and the mean value of the inhibition zones was registered. The results were expressed in terms of inhibition zone (IZ) of growth around each disc in millimeters as: low activity (1–6 mm), moderate activity (7–10 mm), high activity (11–15 mm) and very high activity (12–20 mm) [6].

The minimal inhibition concentrations (MICs) and the minimal bactericidal/fungicidal concentrations (MBCs/MFCs) values were determined for all bacterial/fungal strains by using the microdilution assay as described by Snoussi et al. [7]. The bacterial/fungal inoculums were prepared from 18 h broth cultures and suspensions and spectrophotometrically adjusted to  $10^7$  UFC/mL. The Zn Pc sample was first dissolved in 10% dimethylsulfoxide (DMSO) and then serial twofold dilutions were made in a final concentrations range of 0.0039–8 mg/mL. Firstly, 95  $\mu$ L of nutrient broth was dispersed in each well of plates. An aliquot of 100  $\mu$ L from the stock solutions of the tested Zn Pc was added into the wells. Finally, 5  $\mu$ L from each microbial suspension was added to all the wells. The first well containing 195  $\mu$ L of nutrient broth (Mueller Hinton broth

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or Sabouraud Chloramphenicol broth) without leaves extract and 5  $\mu$ L of the inoculum corresponds to the negative control. The final volume in each well was 200  $\mu$ L. The plates were incubated at 37°C for 24 h.

The MIC was defined as the lowest concentration of the synthetic ZnPc able to inhibit the growth of the microorganisms, so no visible changes were detected in the broth medium. The MBC/MFC ratio corresponds to the highest dilution (lowest concentration) of the sample, which showed clear fluid and an absence of visible growth.

#### Cytotoxicity assay

The cytotoxicity of our compound was evaluated on HEp-2 cells using MTT(dimethyl thiazolyl diphenyl tetrazolium salt) assay as previously described by Hammami and al.[8]. Briefly, 5  $10^4$  Cells were seeded in 96-well culture plates and incubated for 24 hours at 37°C 5% CO<sub>2</sub> in humidified atmosphere to allow cell adhesion and connection. Then, cells were washed twice with PBS and inoculated with 200 µL of our compound diluted in Eagle'S minimal essential medium supplemented with 2% Fetal Calf Serum (two-fold dilutions, ranging from 23, 43 µg to 6000 µg). Cells were incubated for 72 h at 37°C 5% CO<sub>2</sub>. Then, medium was removed, cells were washed in PBS and 40µl of MTT solution (5mg/ml) (MTT, Sigma, Saint Louis, MO, USA) were added to each well. Plates were incubated for 4h at 37°C 5% CO<sub>2</sub>. Metabolic active cells reduce tetrazolium salt to coloured formazan crystals which were solubilized in DMSO (Sigma, USA). The absorbance of formazan generated dye was determined using a microplate reader (Thermo Fisher Scientific, USA) at 540 nm. The percentage of viable cells was calculated using the following formula: % V= (Absorbance of treated cells / Absorbance of control cells), the viability of control cells was set to 100% and the CC<sub>50</sub> value of our compound was calculated from the dose-response curves by linear regression analysis.

#### **RESULTS AND DISCUSSION**

Generally, substituted phthalocyanines are prepared by cyclo-tetramerisation of substituted phthalonitriles. Herein, the synthesis of Zinc phthalocyanine **3** was achieved by treatment of methoxyphthalonitrile **2** with anhydrous zinc (II) acetate in dried butanol and DBU as a catalyst. The proposed structure of Zn(II)-phthalocyanine was confirmed with <sup>1</sup>H, <sup>13</sup>C NMR and mass spectrometry. UV-vis spectra of phthalocyanines are the most reliable signs evidencing the formation of Pc macrocycle. The Q-band absorption is responsible for the characteristically intense green colour of the ZnPc and this band has been interpreted in terms of  $\pi$ - $\pi$ \* excitation between bonding and antibonding molecular orbitals [9].

The electronic spectrum of the ZnPc obtained in dichloromethane solution (Figure 1) shows the characteristic Q-band absorption in the 600-730 nm region. The Soret-band of ZnPc arising from the deeper  $\pi$  levels  $\rightarrow$  LUMO transitions is observed in the UV region at about 300–450 nm.

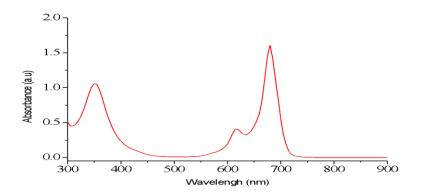


Fig 1: Uv-vis spectrum of ZnPc 3 in a dichloromethane solution (1.5 x 10<sup>-5</sup> M).

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#### **Microbiological test**

Microbiological tests were developed and the results are presented in **table 1**. The results showed that the ZnPc was active against the tested Gram positive and Gram negative bacteria tested with a diameter of growth inhibition zones ranging from 6 to 10 mm. Using the scheme proposed by Parveen and colleagues [6], the ZnPc sample tested can be classified as sample with moderate activity (ZI ranging from 7 to 10 mm). All bacterial strains were particularly sensitive to the ZnPc with low MICs values ranging from 1 to 2 mg/mL. Low concentrations of ZnPc were needed to reproduce a bactericidal effect with a concentration about 4 mg/mL.

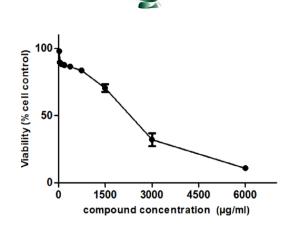
The ZnPc sample was less active against the four *Candida* species tested with a diameter of growth inhibition about 6mm and a MICs values ranging from 1 to 4 mg/ml. The fungicidal effect was obtained from 4 to 8 mg/ml of ZnPc sample.

Microorganisms tested	ZnPc			Ampicillin (10 mg/mL)		
	ZI ± DS	MIC	MBC	ZI ± DS	MIC	MBC
Bacteria						
Staphylococcus aureus ATCC 25923	6.33±0.5 8	1	4	26.66±0.5 7	0.07 8	0.625
Aeromonas hydrophila ATCC 7966 <sup>T</sup>	8±0	2	4	14±0	0.23	0.093
Pseudomonas aeruginosa ATCC 27853	6±0	2	4	22.66±0.5 7	0.11	12
Bacillus cereus ATCC 11778	6±0	1	4	26±1	0.07 8	0.625
Salmonella typhimurium ATCC 1408	7±0	2	4	18±1	0.02 3	0.093
Listeria monocytogenes ATCC 19115	6±0	1	4	12.33±0.5 7	0.02 3	0.093
Escherichia coli ATCC 35218	6±0	1	4	12.33±0.5 7	0.02 3	0.093
Vibrio parahaemolyticus CECT 511	10±0	2	4	17±1	0.02 3	0.093
<u>Yeasts</u>	ZnPc			Amphotericin B (10 mg/mL)		
	ZI±DS	MIC	MBC	ZI±DS	MIC	MFC
Candida glabrata	6±0	2	8	10.33±1.1 5	0.19 5	0.39
Candida albicans	6±0	1	4	11.33±0.5 7	0.01 2	0.024
Candida parapsilosis	6±0	2	8	11±1	0.19 5	0.39
Candida tropicalis	6±0	4	8	7±0	0.19 5	0.39

#### Table 1: Antimicrobial Activity of Zinc-phthalocyanine 3

#### Cytotoxicity assay

The cytotoxicity of our compound was assessed on HEp-2 cells using the MTT salt reduction assay. The cytotoxicity response of the compound was linear and dose-dependent (**Figure 2**). The concentration associated with 50% cytotoxicity was calculated by linear regression analysis and was found to be 2,825 mg/mL. Hence, the tested compound has no toxicity on HEp-2 cells ( $CC_{50} > 1000 \mu g mL^{-1}$ ). [10]



## Fig 2: Cytotoxic effect of the tested compound to HEp-2 cells. The results represent the mean ±SD cell viability percentage of the three independent experiments.

#### CONCLUSION

The study reported in this paper provides a broad range of information about the microbial activities of a Zinc phthalocyanine derivative against four different fungal and nine different bacterial strains. In order to verify the potential risk of ZnPc to human cells, the cytotoxicity of this compound was also assessed.

According to the biological evaluations, all bacterial strains were particularly sensitive to the ZnPc with low MICs values ranging from 1 to 2 mg/mL and less active against the four *Candida* species tested with a diameter of growth inhibition about 6 mm and a MICs values ranging from 1 to 4 mg/mL. The tested compound has no toxicity on HEp-2 cells ( $CC_{50} > 1000 \ \mu g. \ mL^{-1}$ ).

The lack of high cytotoxicity could be beneficial an eventual use as an antimicrobial agent. The synthesized compound could be considered a promising antimicrobial agent and a valuable building block for further derivatization to design more potent biological activity.

CONFLICT OF INTEREST: The authors declare that there is no conflict of interest.

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#### REFERENCES

- [1] Gomis L M, Ohkubo K, LázaroF. F, FukuzumiS, Sastre-SantosÁ. Mater. Chem. A2014;2: 15672-15682
- [2] Aziz F, Ahmad Z,Abdullah S.M,Sulaiman K, SayyadM.H.Pigment & Resin Technology2015;44(1):26-32
- [3] Dhirendra K, ChaudharyP, Kumar k, Lokendra K. Advanced Performance Materials2017;32: 20-28
- [4] IvanM. T, Heidi A.Int J Mol Sci2015;16(5): 10228–10241
- [5] Amel S,Allal L,Mohamed C.Rev Sci Technol2017; 34: 18 -27
- [6] Praveen N, Nayak S, Kar D.M, Das P.J Pharm Res2010;3(6): 1381-1383
- [7] Snoussi M, Noumi E, Trabelsi N, Flamini G, Papetti A, Feo V. De. Mol2015;20: 14402-14424
- [8] Hammami S, Jmii H, El Mokni R, Khmiri A, Faidi K,Dhaouadi H, Aouni M, JoshiR.K. Molecules2015; 20(11): 20426-20433
- [9] Leonardo T. U, Cristiano C. J, Lucilia R. S, Estevão B. P, Samuel M. O, Antonio E. H. M. J. Braz. Chem. Soc 2012; 23(12)
- [10] Prayong P, Barusrux S, Weerapreeyakul N. Fitoterapia2008;79: 598–601.

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